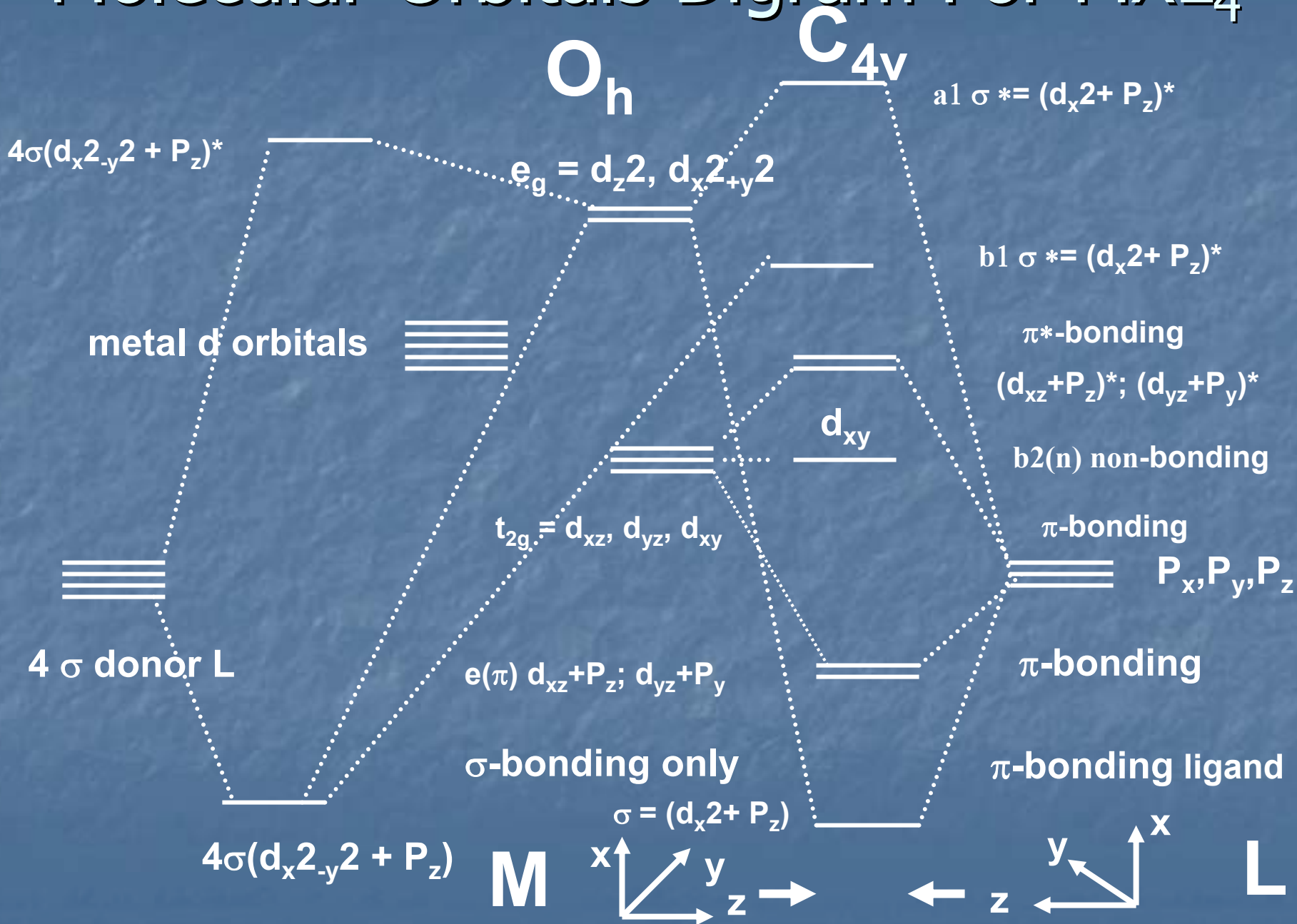


Metal-Ligand Multiple Bonds

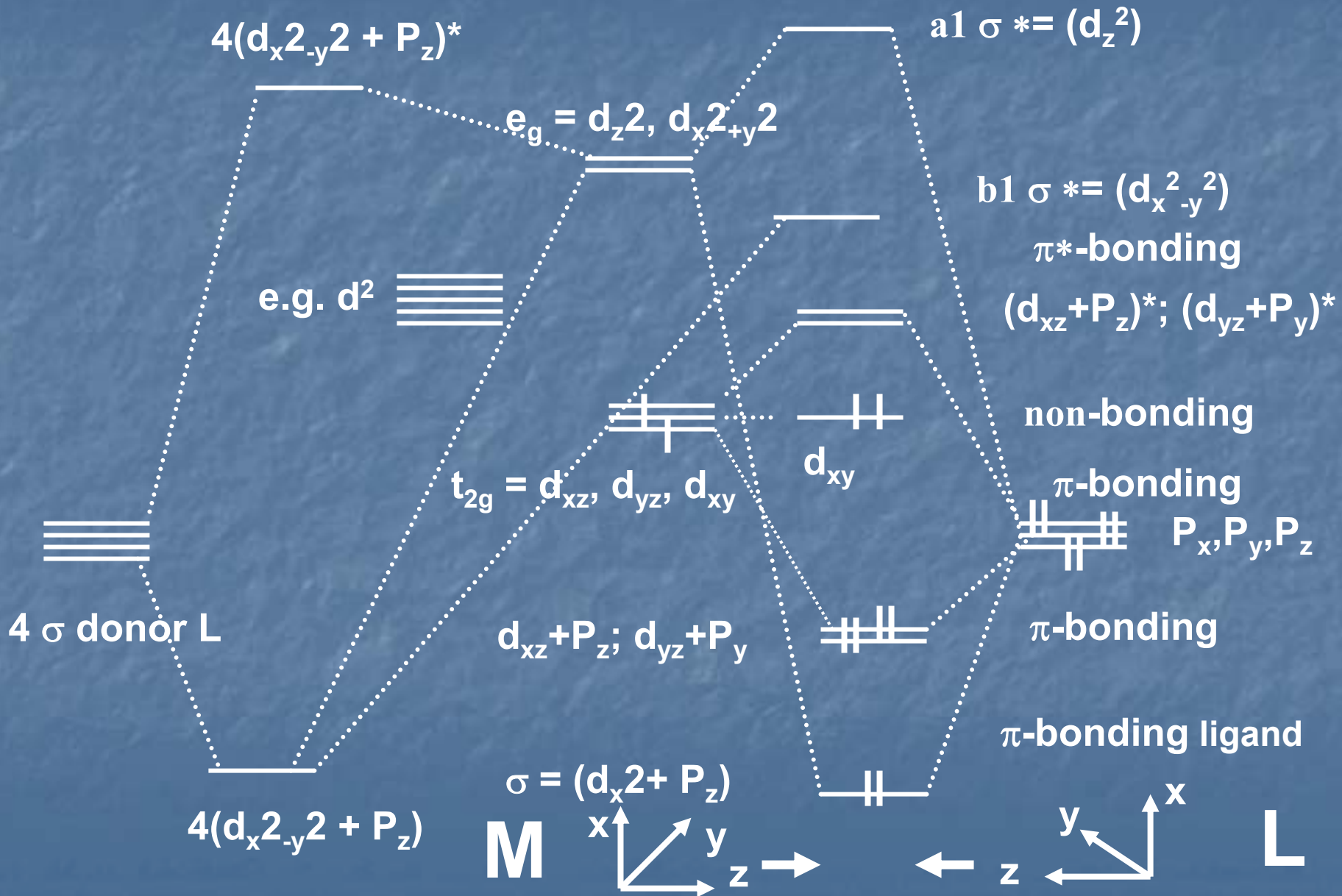
M. Bakir

C60 J

Molecular Orbitals Diagram For MXL_4



Electrons Count



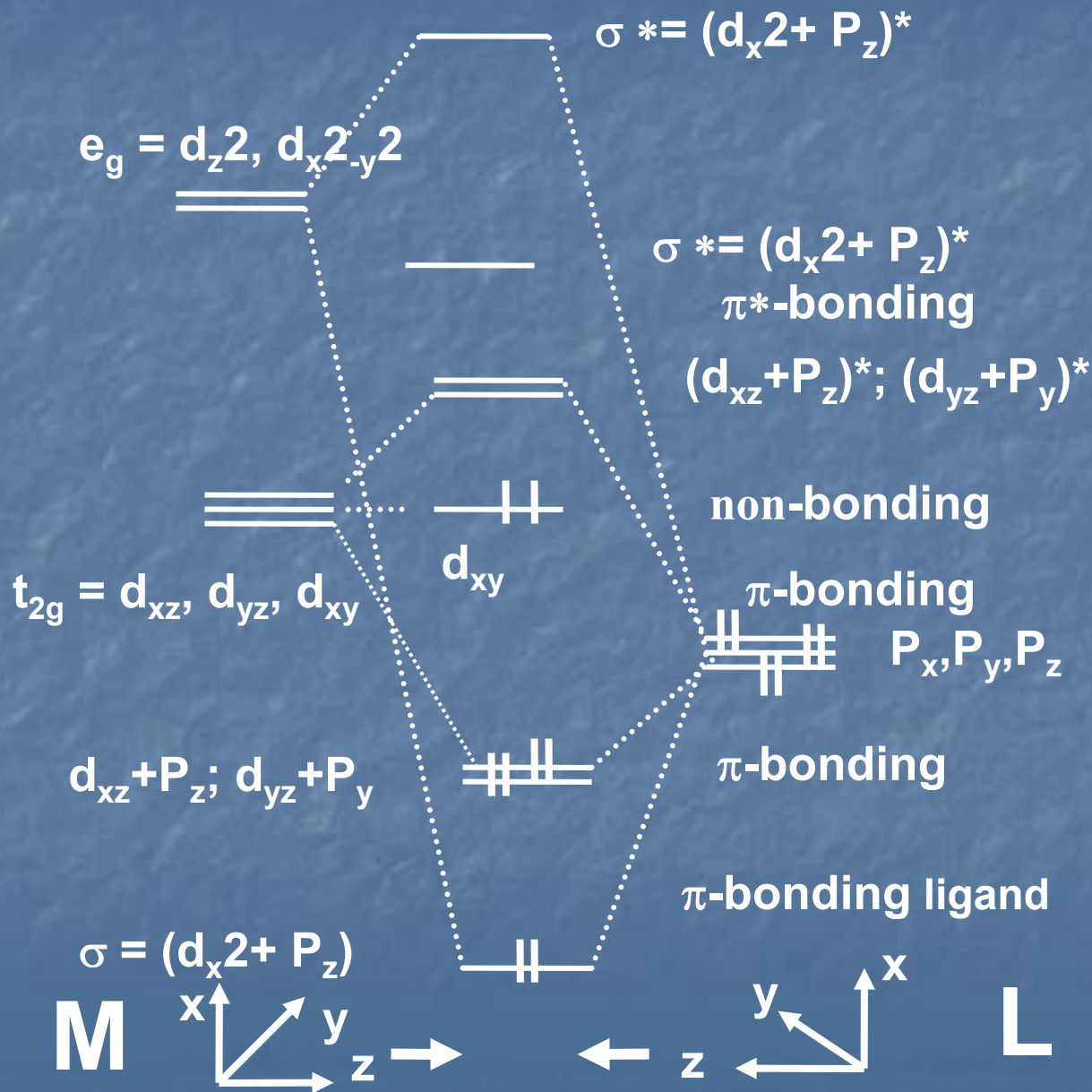
Bond Order

B.O = $\frac{1}{2} (\# \text{ B.E.} - \# \text{ A.E.})$
 In case $n = 0, 1, \text{ or } 2,$
 Where $n = m \# \text{ d e's}$

B.O = 3 Stable

If > 2

Bond order less than 3
 Not stable



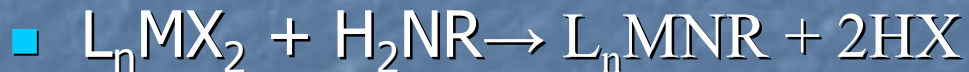
Synthesis of T.M Compounds with Metal-Ligand Multiple Bonds

- Reference: A. A. Cotton & G. Wilkinson; Advanced Inorganic Chemistry, 5th Edition, Wiley
- Metal-Oxides
- Oxidation of metals
 - $M + O_2 \rightarrow MO_n$ (heat)
 - $MX_n + O_2 \rightarrow MO_nX_y$
 - $UCl_4 + O_2 \rightarrow UO_2Cl_2$ at 300 °C
- Reduction of MOn
 - $MO_4 + HX \rightarrow MO_xX_y$
 - $KReO_4 + HX \rightarrow [ReOX_y]^{-n}$ where $y = 4$ or 5 and $n = 1$ or 2
 - $KReO_4 + PPh_3 + HX \rightarrow ReOX_3(PPh_3)_2$

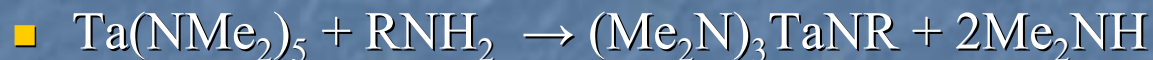
Synthesis of T.M Compounds with Metal-Ligand Multiple Bonds

- Metal Imides

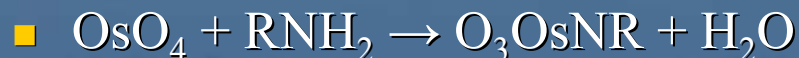
- From Primary Amines



- Metal dialkyamides

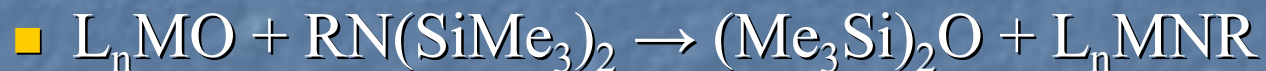


- From Metal oxides

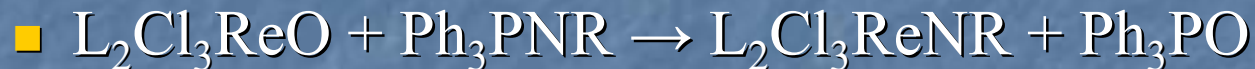
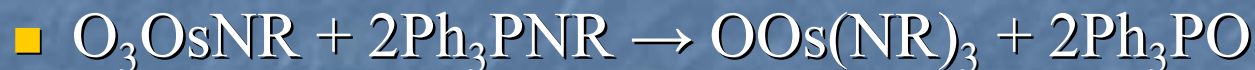


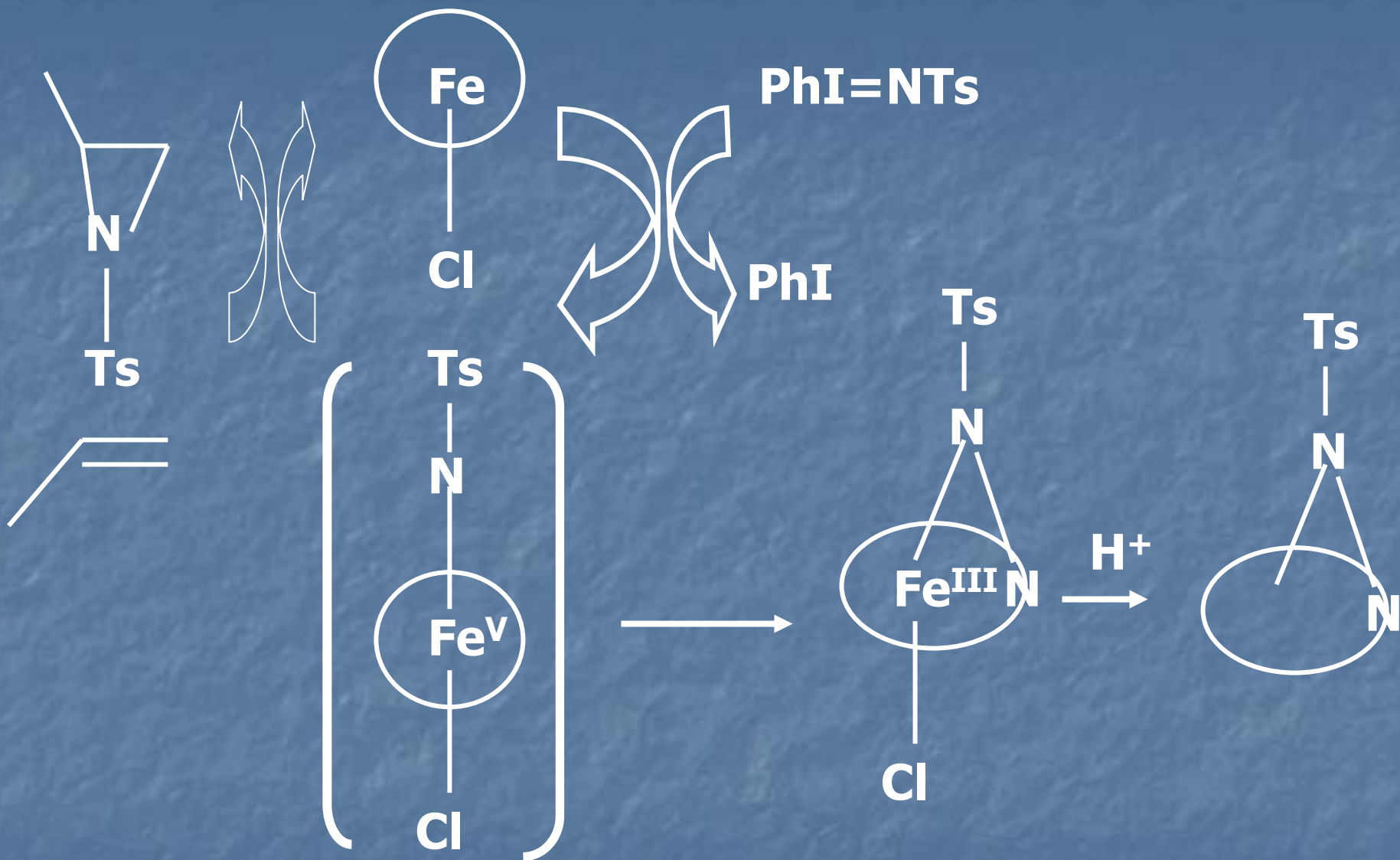
Synthesis of T.M Compounds with Metal-Ligand Multiple Bonds

- From Silylamines



- From phosphinimine, isocyanates and sulfinylamines



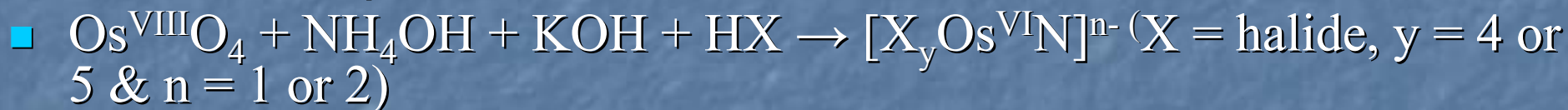


J. P. Mahy, P. Battioni, D. Mansuy, JACS, 1986, 108, 1079.

Synthesis of T.M Compounds with Metal-Ligand Multiple Bonds

- Metal Nitrides

- From Primary Ammonia



- Nitride transfer



- JACS, 2003, 125(44): 13348-13349

- Herdtwick and Herman (2002), J. Organometall. Chem. 660, 121-126.

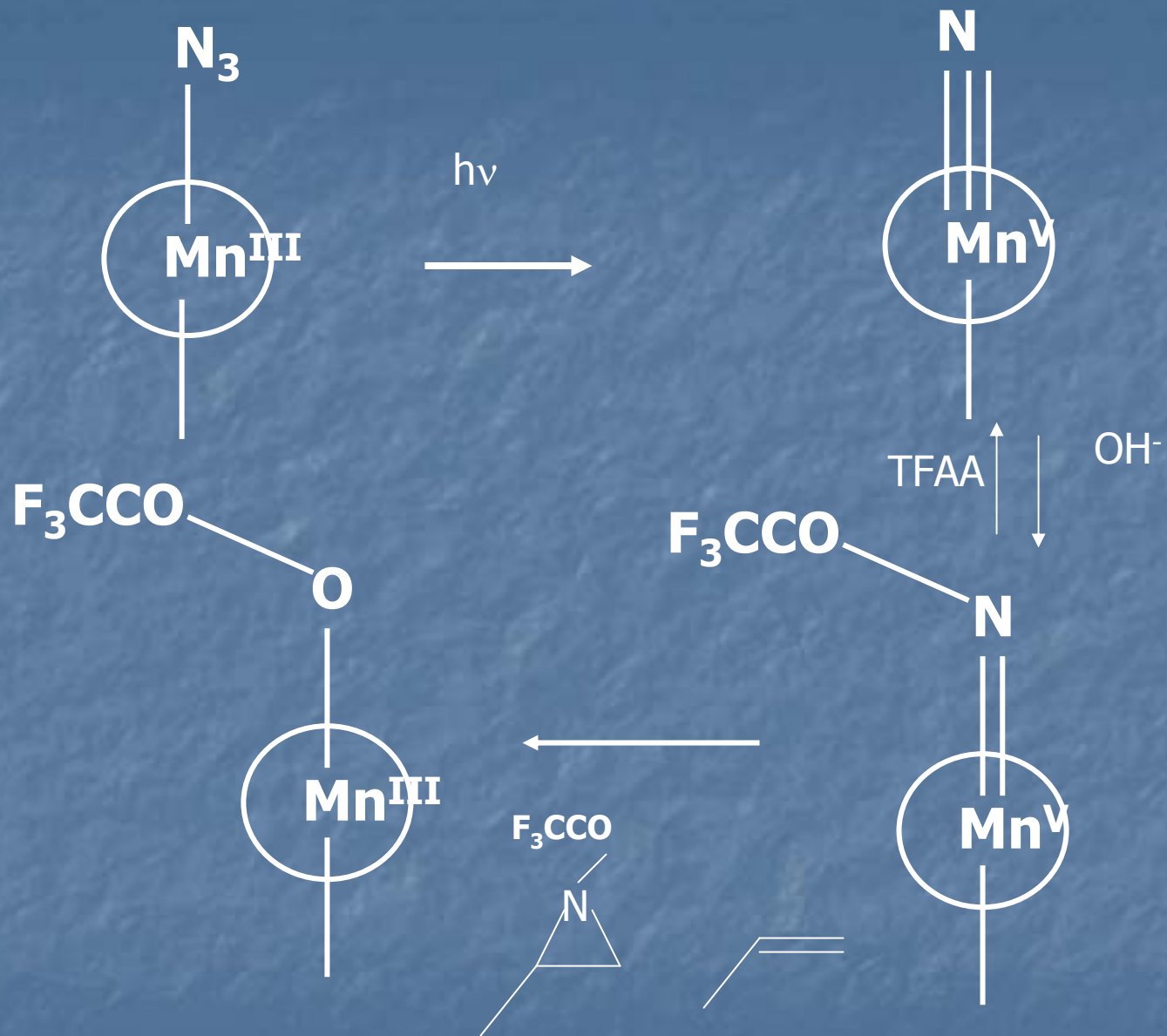
Synthesis of T.M Compounds with Metal-Ligand Multiple Bonds

- Nitrogen splitting



- From hydrazine





J. T. Groves , T. Takahashi, JACS, 1983, 105, 2073